

# HYDROGEN BROMIDE (HBr)

Synonyms : Hydrobromic acid



## Gas Description

Hydrogen Bromide is a colourless, non-flammable gas with a gas density at STP 3.44 kg/m<sup>3</sup>. Hydrogen Bromide fumes in moist air to produce thick acidic clouds. The gas corrosive and is very hygroscopic.

### Typical applications :

Manufacturing of organic and inorganic bromides  
Halogen lamps; cars, photocopiers etc  
Manufacture of pharmaceuticals  
Semiconductors

### Transport Regulations :

UN Number : 1048  
Shipping Name : HYDROGEN BROMIDE, ANHYDROUS  
Class : 2.3 (8) Toxic gases (corrosive)  
CAS Number : 10035-10-6  
Hazchem : 2RE

[Click here](#) for MSDS

[Click here](#) for more information

## Physical Properties

Chemical symbol : HBr  
Molecular weight : 80.92  
Specific gravity (Air=1) : 2.7  
Specific volume (m<sup>3</sup>/kg) : 0.30  
Critical temperature : 90°C  
Boiling point : -66.8°C

Melting Point : NDA  
Critical pressure (bar) : 85.16  
Major hazards : Inhalation, Body Contact  
Toxicity : TWA 3ppm STEL peak limitation  
Flammability Range : Non-Flammable  
Odour : Irritating

[Click here](#) for more information

## Application

### INDUSTRY

#### Electronics

#### Chemicals

### APPLICATION

Hydrogen bromide is used in plasma etching of polysilicon

Used in organic synthesis to add bromine atoms on molecules. The reactive species obtained are intermediates for pharmaceutical products.

[Click here](#) for more information

## Hydrogen Bromide

### Gas

Typical impurities  
(ppm molar)

### Cylinder information

Cylinder Size

Valve Connection

Pressure (kPa)

Contents (m<sup>3</sup>)

Available Upon Request  
Call For Supply Information

### Recommended Equipment

For further information [Contact](#) the specialty gases representative in your region.